Stereoselective Synthesis of Trisubstituted Olefins through 2,5-Dialkylidenemagnesacyclopentanes

V. A. D'yakonov, R. A. Zinnurova, A. G. Ibragimov, and U. M. Dzhemilev

Institute of Petroleum Chemistry and Catalysis, Russian Academy of Sciences, pr. Oktyabrya 141, Ufa, 450075 Bashkortostan, Russia e-mail: ink@anrb.ru

Received September 16, 2006

Abstract—An efficient one-pot procedure for the synthesis of trisubstituted olefins with Z-configured double bonds has been developed on the basis of cross coupling of 2,5-dialkylidenemagnesacyclopentanes (generated in situ) with alkyl, allyl, and aryl halides in the presence of transition metal (Cu, Zr, Fe, Co, Ni, Pd) salts and complexes.

DOI: 10.1134/S1070428007070020

Hydromagnesation of disubstituted acetylenes with isobutylmagnesium bromide [1, 2] or diisobutylmagnesium [3], catalyzed by Cp₂TiCl₂, provides an efficient method for the preparation of (E)-alkenyl-containing organomagnesium reagents. Cross coupling of the latter with organic halogen derivatives opens the way to stereoselective design of trisubstituted olefins that can be used for the preparation of a number of practically important compounds, including those possessing biological activity [4–9]. Available information on the synthesis of (E)-alkenyl organomagensiums from 1,2-dienes is strongly limited [10].

We recently [11] synthesized 2,5-dialkylidenemagnesacyclopentanes by catalytic cyclomagnesation of 1,2-dienes with EtMgCl. The products contain highly reactive vinylic Mg–C bonds and (Z)-alkenyl fragments and are therefore exceptionally promising from the viewpoint of selective synthesis of trisubstituted (1Z,5Z)-diolefins. The goal of the present study was to extend the scope of application of non-Grignard organomagnesium reagents in cross coupling reactions, as well as to develop an efficient one-pot procedure for the synthesis of trisubstituted (Z)-olefins via cross coupling of 2,5-dialkylidenemagnesacyclopentanes with organic halogen derivatives.

The reaction of 2,5-dihexylidenemagnesacyclopentane (Ia) with MeI in the presence of CuCl-2Ph₃P as catalyst was used as model system to determine optimal conditions for the process; at a Ia-to-MeI ratio of 1:2.5 and catalyst concentration of 10 mol % (tetrahydrofuran, $\sim 20^{\circ}$ C), the yield of diolefin IIIa was ~85% ($[M]^+$ 250). The ¹³C NMR spectrum of IIIa



 $R = C_5H_{11}, X = I, R' = Me(a), Et(b), Ph(e); X = Cl, R' = CH_2 = CHCH_2(c), CH_2 = C(Me)CH_2(d); R = C_7H_{15}, R' = Me, X = I(f); R' = C_7H_{15}, R' = Me, X = I(f); R' = ME, X =$ $R' = CH_2 = CHCH_2$, X = Cl(g); $R = PhCH_2$, $R' = CH_2 = CHCH_2$, X = Cl(h); $[Ti] = Cp_2TiCl_2$; [M] = Cu, Pd, Ni, Co.

contained upfield signals from C^5/C^{12} at δ_C 28.07 ppm and C^8/C^9 at δ_C 29.86 ppm, which are typical of carbon atoms in the allylic position with respect to *cis*-configured double bonds in an alkadiene fragment. The double-bonded C^6/C^{11} and C^7/C^{10} atoms resonated at δ_C 126.56 and 137.44 ppm, respectively. These data allowed us to assigned the structure of (6*Z*,10*Z*)-7,10dimethylhexadeca-6,10-diene to hydrocarbon **IIIa**.

We also tested various transition metal salts and complexes (Cu, Zr, Fe, Co, Ni, Pd) as catalysts in the cross coupling of magnesacyclopentane **Ia** with MeI. As a result, the following activity series was obtained: CuCl + 2Ph₃P (85%) > CuCl (84%) > Pd(acac)₂ + 2Ph₃P (83%) > Cu(acac)₂ + 2Ph₃P (82%) > CuI (81%) > PdCl₂ + 2Ph₃P (72%) > Ni(acac)₂ + 2Ph₃P (69%) > Pd(Ph₃P)₄ (68%) > Fe(acac)₃ + 2Ph₃P (69%) > Pd(Ph₃P)₂Cl₂ (55%) > Pd(dba)₂ + 2Ph₃P (57%) > NiCl₂ + 2Ph₃P (32%) > CoCl₂ + 2Ph₃P (28%). It is seen that Cu, Pd, and Ni compounds ensure the highest yields of target cross-coupling product **IIIa**.

The reaction of 2,5-dihexylidenemagnesacyclopentane (**Ia**) with an equimolar amount of methyl iodide gave ~64% of (6*Z*,10*Z*)-7-methylhexadeca-6,10-diene (**IIa**) having di- and trisubstituted double bonds with *cis* configuration. This follows from the presence of upfield signals in the ¹³C NMR spectrum at $\delta_{\rm C}$ 28.05 and 29.18 ppm for C⁵ and C⁸ and at $\delta_{\rm C}$ 26.97 and 27.42 ppm for C⁹ and C¹², respectively. Apart from compound **IIa**, ~10% of (6*Z*,10*Z*)-7,10-dimethylhexadeca-6,10-diene (**IIIa**) was formed via cross coupling of magnesacyclopentadiene **Ia** with two molecules of methyl iodide.

Under the optimal conditions (10 mol % of CuCl + 2Ph₃P, THF, 20°C, 8 h) we performed reactions of 2,5-dialkylidenemagnesacyclopentanes **Ib–Ih** with ethyl and phenyl iodides and allyl and 2-methylprop-2en-1-yl chlorides and obtained the corresponding products **IIb–IIh** or **IIIb–IIIh** in 75–85% yield (Scheme 1). The ratio of compounds **II** and **III** depended on the ratio of initial 2,5-dialkylidenemagnesacyclopentanes **I** and organic halogen derivatives.

Our results demonstrate that transition-metal catalyzed cross coupling of 2,5-dialkylidenemagnesacyclopentanes with halogen derivatives provides an efficient method for the stereoselective synthesis of di- and trisubstituted diolefins in high yields.

EXPERIMENTAL

The hydrolysis products were analyzed by GLC on a Chrom-5 chromatograph (carrier gas helium; $1200 \times$ 3-mm column packed with 5% of SE-30 or 15% of PEG-6000 on Chromaton N-AW). The IR spectra were recorded on a Specord 75IR spectrometer from thin films (neat). The ¹H and ¹³C NMR spectra were measured from solutions in CDCl₃ on Jeol FX-90Q (89.55 MHz for ¹H and 22.5 MHz for ¹³C) and Bruker AM-300 spectrometers (300 MHz for ¹H and 75.46 MHz for ¹³C). The mass spectra (electron impact, 70 eV) were obtained on an MKh-1306 instrument (ion source temperature 200°C). All experiments were carried out under dry argon. Solutions of ethylmagnesium bromide in THF were prepared by the procedure described in [12]; THF was heated under reflux over metallic sodium and then distilled under argon prior to use. Commercial Cp₂TiCl₂ was used. Initial 1,2-dienes were synthesized according to [13]. Reactions with organometallic compounds were carried out in a stream of dry argon. The yields were determined by GLC analysis of the hydrolysis products.

Reaction of 2,5-dialkylidenemagnesacyclopentanes with organic halides (general procedure). A glass reactor was charged under dry argon at 0°C with 12 mmol of a freshly prepared solution of EtMgCl in THF, 6 mmol of magnesium powder, 10 mmol of the corresponding 1,2-diene, and 0.5 mmol of Cp₂TiCl₂. The mixture was allowed to warm up to room temperature, stirred for 8 h on a magnetic stirrer, and cooled to -20°C, 1 mmol of CuCl-2Ph₃P was added, 5 or 12.5 mmol of the corresponding halogen derivative was added dropwise, and the mixture was allowed to warm up to room temperature and stirred for 6 h. The mixture was then treated with 5% hydrochloric acid and extracted with diethyl ether, the extract was dried over MgSO₄, the solvent was removed, and compounds IIa-IIh and IIIa-IIIh were isolated by vacuum distillation.

(6Z,10Z)-7-Methylhexadeca-6,10-diene (IIa). bp 110–111°C (1 mm). IR spectrum, v, cm⁻¹: 3050, 2910, 2810, 1640, 1490, 1460, 1000, 920, 890, 710, 700. ¹H NMR spectrum, δ , ppm: 0.65–0.89 m (6H, CH₃), 1.11–1.35 m (12H, CH₂), 1.61 s (3H, CH₃), 1.81–2.21 m (8H, CH₂C=C), 4.86–4.97 m (1H, HC=C), 5.0–5.36 m (2H, HC=C). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 14.05 (C¹), 14.16 (C¹⁶), 22.51 (C²), 22.61 (C¹⁵), 22.63 (CH₃), 26.97 (C⁹), 27.42 (C¹²), 28.05 (C⁵), 29.18 (C⁸), 29.27 (C⁴), 29.89 (C¹³), 31.45 (C³), 31.56 (C¹⁴), 126.56 (C⁶), 128.97 (C¹⁰), 130.75 (C¹¹), 136.46 (C⁷). Found, %: C 85.01; H 13.15. [*M*]⁺ 236. C₁₇H₃₂. Calculated, %: C 86.36; H 13.64.

(6Z,10Z)-7,10-Dimethylhexadeca-6,10-diene (IIIa). bp $121-122^{\circ}C$ (1 mm). IR spectrum, v, cm⁻¹: 3030, 2900, 2800, 1650, 1490, 1450, 1000, 910, 890, 720, 700. ¹H NMR spectrum, δ , ppm: 0.62–0.87 t (6H, CH₃, *J* = 6.0 Hz), 1.10–1.36 m (12H, CH₂), 1.72 s (6H, CH₃), 1.80–2.22 m (8H, CH₂C=C), 4.88–4.95 m (2H, CH=C). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 13.96 (C¹, C¹⁶), 22.52 (C², C¹⁵), 22.63 (CH₃), 28.07 (C⁵, C¹²), 29.75 (C⁴, C¹³), 29.86 (C⁸, C⁹), 31.60 (C³, C¹⁴), 126.56 (C⁶, C¹¹), 137.44 (C⁷, C¹⁰). Found, %: C 84.89; H 12.95. [*M*]⁺ 250. C₁₈H₃₄. Calculated, %: C 86.32; H 13.68.

(6Z,10Z)-7-Ethylhexadeca-6,10-diene (IIb). bp 120–122°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2900, 2800, 1650, 1495, 1460, 1000, 910, 895, 720, 695. ¹H NMR spectrum, δ , ppm: 0.65–1.05 m (9H, CH₃), 1.13–1.39 m (12H, CH₂), 1.81–2.21 m (10H, CH₂C=C), 4.84–4.95 m (1H, HC=C), 5.05–5.34 m (2H, HC=C). ¹³C NMR spectrum, δ_{C} , ppm: 14.05 (C¹), 13.02 (CH₃), 14.12 (C¹⁶), 22.53 (C²), 22.60 (C¹⁵), 25.81 (C⁹), 26.48 (CH₂), 27.85 (C¹²), 28.11 (C⁵), 29.19 (C⁸), 29.31 (C⁴), 29.79 (C¹³), 31.42 (C³), 31.54 (C¹⁴), 126.51 (C⁶), 128.87 (C¹⁰), 130.74 (C¹¹), 137.46 (C⁷). Found, %: C 85.75; H 12.99. [*M*]⁺ 250. C₁₈H₃₄. Calculated, %: C 86.32; H 13.68.

(6Z,10Z)-7,10-Diethylhexadeca-6,10-diene (IIIb). bp 140–142°C (1 mm). IR spectrum, v, cm⁻¹: 3050, 2900, 2810, 1650, 1480, 1450, 1000, 920, 890, 725, 700. ¹H NMR spectrum, δ , ppm: 0.62–0.99 m (12H, CH₃), 1.12–1.38 m (12H, CH₂), 1.81–2.25 m (12H, CH₂C=C), 4.80–4.91 m (2H, CH=C). ¹³C NMR spectrum, δ_{C} , ppm: 12.58 (CH₃), 14.02 (C¹, C¹⁶), 22.55 (C², C¹⁵), 26.52 (CH₂), 28.17 (C⁵, C¹²), 29.74 (C⁴, C¹³), 29.89 (C⁸, C⁹), 31.63 (C³, C¹⁴), 126.46 (C⁶, C¹¹), 138.64 (C⁷, C¹⁰). Found, %: C 84.75; H 13.06. [*M*]⁺ 278. C₂₀H₃₈. Calculated, %: C 86.25; H 13.75.

(6*E*,10*Z*)-7-Allylhexadeca-6,10-diene (IIc). bp 133–134°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2900, 2870, 1640, 1490, 1460, 1380, 1000, 920, 890, 720. ¹H NMR spectrum, δ , ppm: 0.72–0.91 m (6H, CH₃), 1.21–1.45 m (12H, CH₂), 1.89–2.21 m (8H, CH₂C=C), 2.52–2.74 d (2H, C=CCH₂C=C), 4.83– 4.90 m (2H, H₂C=C), 5.0–5.14 t (1H, HC=C, *J* = 6.0 Hz), 5.26–5.38 m (2H, HC=CH), 5.58–5.86 m (1H, H₂C=CH). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 14.05 (C¹), 14.09 (C¹⁶), 22.54 (C²), 22.62 (C¹⁵), 25.99 (C⁹), 27.84 (C¹²), 28.15 (C⁵), 29.28 (C⁸), 29.37 (C⁴), 29.98 (C¹³), 31.46 (C³), 31.46 (C¹⁴), 41.86 (CH₂), 115.49 (=CH₂), 126.61 (C⁶), 129.34 (C¹⁰), 130.75 (C¹¹), 137.26 (C⁷), 137.50 (CH=). Found, %: C 86.12; H 12.55. [*M*]⁺ 262. C₁₉H₃₄. Calculated, %: C 86.94; H 13.06.

(6*E*,10*E*)-7,10-Diallylhexadeca-6,10-diene (IIIc). bp $164-166^{\circ}C$ (1 mm). IR spectrum, v, cm⁻¹: 3030, 2950, 2860, 1650, 1490, 1450, 1390, 1000, 910, 890, 720. ¹H NMR spectrum, δ , ppm: 0.75–0.92 t (6H, CH₃, J = 6.0 Hz), 1.21–1.42 m (12H, CH₂), 1.91–2.25 m (8H, CH₂C=C), 2.55–2.75 d (4H, C=CCH₂C=C), 4.85– 4.96 m (4H, H₂C=C), 5.0–5.15 t (2H, CH₂CH=C, J =6.0 Hz), 5.55–5.85 m (2H, H₂C=CH). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 13.92 (C¹, C¹⁶), 22.54 (C², C¹⁵), 29.0 (C⁴, C¹³), 29.20 (C⁵, C¹²), 29.72 (C⁸, C⁹), 29.75 (C⁴, C¹³), 31.64 (C³, C¹⁴), 41.84 (CH₂), 115.52 (=CH₂), 126.56 (C⁶, C¹¹), 136.44 (C⁷, C¹⁰), 137.50 (CH=). Found, %: C 86.25; H 11.81. [*M*]⁺ 302. C₂₂H₃₈. Calculated, %: C 87.34; H 12.66.

(6E,10Z)-7-(2-Methylprop-2-en-1-yl)hexadeca-6,10-diene (IId). bp 144–145°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2900, 2860, 1640, 1480, 1460, 1390, 1000, 910, 890, 720. ¹H NMR spectrum, δ, ppm: 0.79– 0.91 m (6H, CH₃), 1.21–1.45 m (12H, CH₂), 1.63 s (3H, CH₃), 1.95–2.21 m (8H, CH₂C=C), 2.69–2.74 d (2H, C=CCH₂C=C), 4.8 s (2H, H₂C=C), 5.0–5.14 t $(1H, CH_2CH=C, J = 6.0 Hz), 5.22-5.38 m (2H, CH_2CH=C)$ CH=CH). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 14.03 (C¹), 14.09 (C^{16}), 20.21 (CH_3), 22.52 (C^2), 22.59 (C^{15}), 25.94 (C⁹), 27.79 (C¹²), 28.25 (C⁵), 29.29 (C⁸), 29.47 (C⁴), 29.88 (C¹³), 31.36 (C³), 31.41 (C¹⁴), 46.49 (CH₂), 111.82 (=CH₂), 126.69 (C^6), 129.24 (C^{10}), 130.66 (C^{11}) , 137.21 (C^{7}) , 144.41 $(CH_{3}C=)$. Found, %: C 85.14; H 12.75. [*M*]⁺ 276. C₂₀H₃₆. Calculated, %: C 86.88; H 13.12.

(6*E*,10*E*)-7,10-Bis(2-methylprop-2-en-1-yl)hexadeca-6,10-diene (IIId). bp 187–188°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2950, 2860, 1650, 1480, 1450, 1390, 1000, 920, 890, 720. ¹H NMR spectrum, δ, ppm: 0.81–0.92 t (6H, CH₃, *J* = 6.0 Hz), 1.21– 1.35 m (12H, CH₂), 1.65 s (6H, CH₃), 2.05–2.20 m (8H, CH₂C=C), 2.75–2.85 m (4H, C=CCH₂C=C), 4.85 s (4H, H₂C=C), 5.25–5.45 m (2H, C=CH). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 14.06 (C¹, C¹⁶), 20.24 (CH₃), 22.62 (C², C¹⁵), 27.96 (C⁵, C¹²), 28.45 (C⁸, C⁹), 29.78 (C⁴, C¹³), 31.67 (C³, C¹⁴), 46.52 (CH₂), 111.84 (=CH₂), 127.70 (C⁶, C¹¹), 136.88 (C⁷, C¹⁰), 144.53 (CH₃C=). Found, %: C 86.46; H 12.11. [*M*]⁺ 302. C₂₄H₄₂. Calculated, %: C 87.19; H 12.81.

(6*E*,10*Z*)-7-Phenylhexadeca-6,10-diene (IIe). bp 172–174°C (1 mm). IR spectrum, v, cm⁻¹: 3050, 2950, 2860, 1640, 1490, 1450, 1390, 1000, 920, 890, 710. ¹H NMR spectrum, δ , ppm: 0.76–0.94 m (6H, CH₃), 1.23–1.94 m (12H, CH₂), 1.95–2.22 m (8H, CH₂C=C), 5.0–5.37 m (2H, CH=CH), 5.55–5.85 m (1H, C=CH), 7.0–7.5 m (5H, Ph). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 14.03 (CH₃), 14.12 (C¹⁰), 22.58 (CH₂), 22.61 (C⁹), 27.61 (C³), 27.82 (C⁶), 29.54 (CH₂), 30.35 (C⁷), 30.63 (CH₂), 31.51 (C⁸), 31.58 (CH₂), 32.12 (C²), 125.23 (C_{arom}), 126.35 (CH=), 126.64 (C_{arom}), 128.09 (C⁴), 128.16 (C¹), 128.37 (C_{arom}), 130.62 (C⁵), 145.34 (C_{arom}). Found, %: C 87.67; H 10.93. [*M*]⁺ 298. C₂₂H₃₄. Calculated, %: C 88.52; H 11.48.

(6*E*,10*Z*)-7,10-Diphenylhexadeca-6,10-diene (IIIe). bp 248–249°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2950, 2860, 1650, 1490, 1450, 1390, 1000, 910, 890, 720. ¹H NMR spectrum, δ , ppm: 0.63–0.88 t (6H, CH₃, *J* = 6.0 Hz), 1.12–1.36 m (12H, CH₂), 1.80– 2.24 m (8H, CH₂C=C), 4.86–4.95 m (2H, CH=C), 7.0– 7.55 m (10H, Ph). ¹³C NMR spectrum, δ_{C} , ppm: 13.98 (C¹, C¹⁶), 22.58 (C², C¹⁵), 29.56 (C⁴, C¹³), 30.25 (C⁵, C¹²), 30.46 (C⁸, C⁹), 31.60 (C³, C¹⁴), 125.15 (C_{arom}), 126.58 (C⁶, C¹¹), 126.62 (C_{arom}), 127.42 (C⁷, C¹⁰), 128.47 (C_{arom}), 142.54 (C_{arom}). Found, %: C 88.36; H 9.86. [*M*]⁺ 374. C₂₈H₃₈. Calculated, %: C 89.78; H 10.22.

(8Z,12Z)-9-Methylicosa-8,12-diene (IIf). bp 155– 157°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2920, 2800, 1650, 1480, 1460, 1000, 910, 890, 720, 700. ¹H NMR spectrum, δ , ppm: 0.65–0.89 m (6H, CH₃), 1.13–1.35 m (20H, CH₂), 1.62 s (3H, CH₃), 1.81– 2.25 m (8H, CH₂C=C), 4.87–4.94 m (1H, HC=C), 5.1– 5.46 m (2H, HC=C). ¹³C NMR spectrum, δ_{C} , ppm: 14.05 (C¹, C²⁰); 22.71 (C², C¹⁹); 22.65 (CH₃); 26.97 (C¹¹); 27.67 (C⁷); 27.82 (C¹⁴); 28.05 (C¹⁰); 29.33, 29.42, 29.81, 29.97 (C⁵, C⁶, C¹⁶, C¹⁷); 30.21 (C¹⁵); 30.34 (C³, C¹⁸); 31.46 (C⁴); 125.86 (C⁸); 129.87 (C¹³); 130.24 (C¹²); 138.15 (C⁹). Found, %: C 85.73; H 13.06. [*M*]⁺ 292. C₂₁H₄₀. Calculated, %: C 86.22; H 13.78.

(8Z,12Z)-9,12-Dimethylicosa-8,12-diene (IIIf). bp 166–168°C (1 mm). IR spectrum, v, cm⁻¹: 3050, 2910, 2820, 1650, 1480, 1450, 1050, 920, 890, 710, 700. ¹H NMR spectrum, δ , ppm: 0.65–0.89 t (6H, CH₃, J = 6.0 Hz), 1.15–1.35 m (20H, CH₂), 1.65 s (6H, CH₃), 1.85–2.23 m (8H, CH₂C=C), 4.89–4.97 m (2H, CH=C). ¹³C NMR spectrum, δ_{C} , ppm: 14.05 (C¹, C²⁰), 22.57 (C², C¹⁹), 22.64 (CH₃), 28.07 (C⁶, C¹⁵), 29.64 and 29.75 (C⁴, C⁵, C¹⁶, C¹⁷), 29.89 (C¹⁰, C¹¹), 30.09 (C⁷, C¹⁴), 31.62 (C³, C¹⁸), 126.66 (C⁸, C¹³), 137.45 (C⁹, C¹²). Found, %: C 85.92; H 13.05. [*M*]⁺ 306. C₂₂H₄₂. Calculated, %: C 86.19; H 13.81.

(8*E*,12*Z*)-9-Allylicosa-8,12-diene (IIg). bp 176– 177°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2950, 2870, 1640, 1490, 1460, 1380, 1000, 910, 890, 720. ¹H NMR spectrum, δ , ppm: 0.75–0.92 m (6H, CH₃), 1.21–1.42 m (20H, CH₂), 1.89–2.21 m (8H, CH₂C=C), 2.55–2.74 d (2H, C=CCH₂C=C), 4.81–4.90 m (2H, H₂C=C), 5.0–5.14 t (1H, HC=C, J = 6.0 Hz), 5.25– 5.36 m (2H, HC=CH), 5.58–5.84 m (1H, H₂C=CH). ¹³C NMR spectrum, δ_C, ppm: 14.06 (C¹, C²⁰); 22.52 (C², C¹⁹); 26.27 (C¹¹); 27.85 (C¹⁴); 28.13 (C⁶); 29.21 (C⁷); 29.27 (C¹⁰); 29.37, 29.46, 29.56, 29.98 (C⁴, C⁵, C¹⁶, C¹⁷); 30.14 (C¹⁵); 31.42 (C³, C¹⁸); 41.74 (CH₂); 115.36 (=CH₂); 126.72 (C⁸); 129.41 (C¹²); 130.65 (C¹³); 137.16 (C⁹); 137.52 (CH=). Found, %: C 86.15; H 12.78. [M]⁺ 262. C₂₃H₄₂. Calculated, %: C 86.71; H 13.29.

(8*E*,12*E*)-9,12-Diallylicosa-8,12-diene (IIIg). bp 207–208°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2960, 2860, 1650, 1490, 1450, 1390, 1050, 920, 890, 720. ¹H NMR spectrum, δ , ppm: 0.74–0.91 m (6H, CH₃), 1.22–1.44 m (20H, CH₂), 1.90–2.23 m (8H, CH₂C=C), 2.56–2.75 d (4H, C=CCH₂C=C), 4.84– 4.96 m (4H, H₂C=C), 5.05–5.15 t (2H, HC=C, *J* = 6.0 Hz), 5.56–5.84 m (2H, H₂C=CH). ¹³C NMR spectrum, δ_{C} , ppm: 14.01 (C¹, C²⁰), 22.56 (C², C¹⁹), 28.16 (C⁶, C¹⁵), 29.06 and 29.26 (C⁴, C⁵, C¹⁶, C¹⁷), 29.28 (C⁷, C¹⁴), 29.78 (C¹⁰, C¹¹), 31.67 (C³, C¹⁸), 41.91 (CH₂), 115.48 (=CH₂), 126.58 (C⁸, C¹³), 136.34 (C⁹, C¹²), 137.56 (CH=). Found, %: C 86.52; H 12.31. [*M*]⁺ 358. C₂₆H₄₆. Calculated, %: C 87.07; H 12.93.

(2*E*,6*Z*)-3-Allyl-1,8-diphenylocta-2,6-diene (IIh). bp 200–202°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2920, 2800, 1650, 1490, 1460, 1000, 920, 890, 720, 700. ¹H NMR spectrum, δ , ppm: 1.90–2.25 m (4H, CH₂C=C), 2.55–2.76 d (2H, C=CCH₂C=C), 3.35– 3.48 m (4H, CH₂Ph), 4.85–4.96 m (2H, H₂C=C), 5.0– 5.15 m (3H, HC=CH, CH=C), 5.55–5.86 m (1H, H₂C=CH), 7.11–7.36 m (10H, Ph). ¹³C NMR spectrum, δ_{C} , ppm: 29.69 (C⁴, C⁵), 33.48 (C⁸), 33.89 (C¹), 41.59 (CH₂), 115.82 (=CH₂), 125.41 (C_{arom}), 127.50 (C²), 127.74 (C⁷), 128.14 (C_{arom}), 129.08 (C_{arom}), 130.71 (C⁶), 136.69 (=CH), 138.20 (C³), 141.17 (C_{arom}). Found, %: C 90.98; H 8.05. [*M*]⁺ 302. C₂₃H₂₆. Calculated, %: C 91.34; H 8.66.

(2*E*,6*E*)-3,6-Diallyl-1,8-diphenylocta-2,6-diene (IIIh). bp 180–181°C (1 mm). IR spectrum, v, cm⁻¹: 3030, 2900, 2800, 1650, 1490, 1450, 1000, 910, 890, 720, 700. ¹H NMR spectrum, δ , ppm: 1.91–2.25 m (4H, CH₂CH=C), 2.55–2.75 d (4H, C=CHCH₂C=C), 3.32–3.41 d (4H, CH₂Ph), 4.85–4.96 m (4H, H₂C=C), 5.0–5.15 t (2H, HC=C, *J* = 6.0 Hz), 5.55–5.85 m (2H, H₂C=CH), 7.15–7.35 m (10H, Ph). ¹³C NMR spectrum, δ_{C} , ppm: 28.81 (C⁴, C⁵), 33.9 (C¹, CH₂Ph), 41.61 (C⁷, CH₂), 115.98 (C⁹, =CH₂), 125.68 (C_{arom}), 127.56 (C², =CH), 128.19 (C_{arom}), 129.11 (C_{arom}), 136.8 (C⁸, =CH), 138.35 (C³, C⁶), 141.18 (C_{arom}). Found, %: C 91.01; H 8.12. $[M]^+$ 342. $C_{26}H_{30}$. Calculated, %: C 91.17; H 8.83.

The authors thank L.M. Khalilov (Spectral Laboratory, Institute of Petroleum Chemistry and Catalysis, Russian Academy of Sciences) for helpful discussion of the NMR spectra. This study was performed under financial support by the Russian Foundation for Basic Research (project no. 05-03-32367, NSh-7470.2006.3) and by the Foundation for Support of Russian Science.

REFERENCES

- 1. Sato, F., Ishikawa, H., and Sato, M., *Tetrahedron Lett.*, 1981, vol. 22, p. 85.
- 2. Sato, F., J. Organomet. Chem., 1985, vol. 285, p. 53.
- Dzhemilev, U.M., Vostrikova, O.S., Sultanov, R.M., and Gimaeva, A.R., *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1988, p. 2156.
- 4. Sato, F., Ishikawa, H., Watanabe, H., Miyake, T., and Sato, M., J. Chem. Soc., Chem. Commun., 1981, p. 718.
- Sato, F. and Katsuno, H., *Tetrahedron Lett.*, 1983, vol. 24, p. 1809.

- Sato, F., Watanabe, H., Tonaka, Y., Yamaji, T., and Sato, M., *Tetrahedron Lett.*, 1983, vol. 24, p. 1041.
- 7. Sato, F., Kusakabe, M., and Kobayashi, Y., J. Chem. Soc., Chem. Commun., 1984, p. 1130.
- Kocienski, P., Love, C., and Whitby, R., *Tetrahedron Lett.*, 1988, vol. 29, p. 2867.
- 9. Jefford, C.W. and Moulin, M.C., *Helv. Chim. Acta*, 1991, vol. 74, no. 2, p. 336.
- Marek, I. and Normant, J.F., *Metal-Catalyzed Cross-Coupling Reactions*, Diederich, F. and Stang, P.J., Eds., New York: Wiley, 1998, p. 271.
- Dzhemilev, U.M., D'yakonov, V.A., Khafizova, L.O., and Ibragimov, A.G., *Tetrahedron*, 2004, vol. 60, p. 1287.
- Ioffe, S.T. and Nesmeyanov, A.N., Metody elementoorganicheskoi khimii. Podgruppa magniya, berilliya, kal'tsiya, strontsiya, bariya (Methods of Organometallic Chemistry. Magnesium, Beryllium, Calcium, Strontium, and Barium Subgroup), Moscow: Akad. Nauk SSSR, 1963.
- 13. Sato, F., Oguro, K., and Sato, M., *Chem. Lett.*, 1978, vol. 7, p. 805.